

# Chemical Reactions



Topic 7

# Reactions

- Chemical reactions have two or more substances undergoing a reorganization of atoms
- These reactions always form a new substance
- The reactants turn into a new set of products



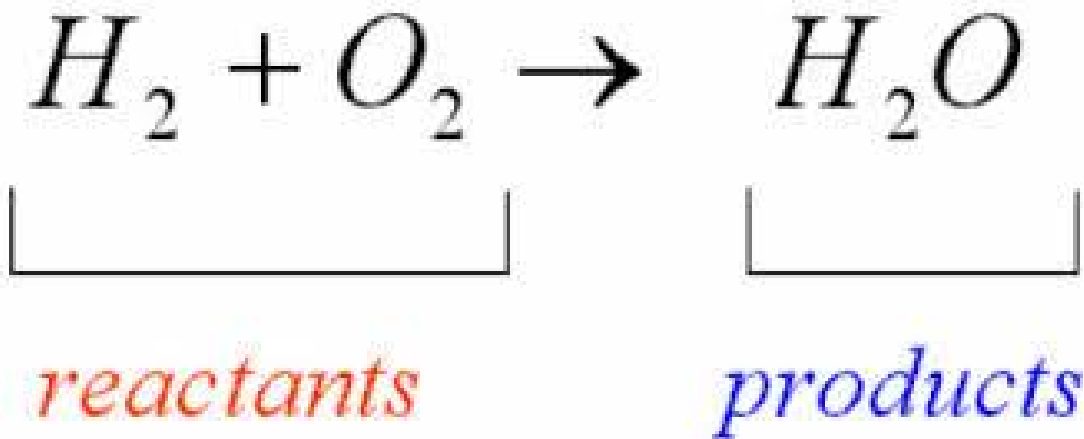
# Chemical Equations

Ex.

Silver + Bromine  $\Rightarrow$  Silver bromide

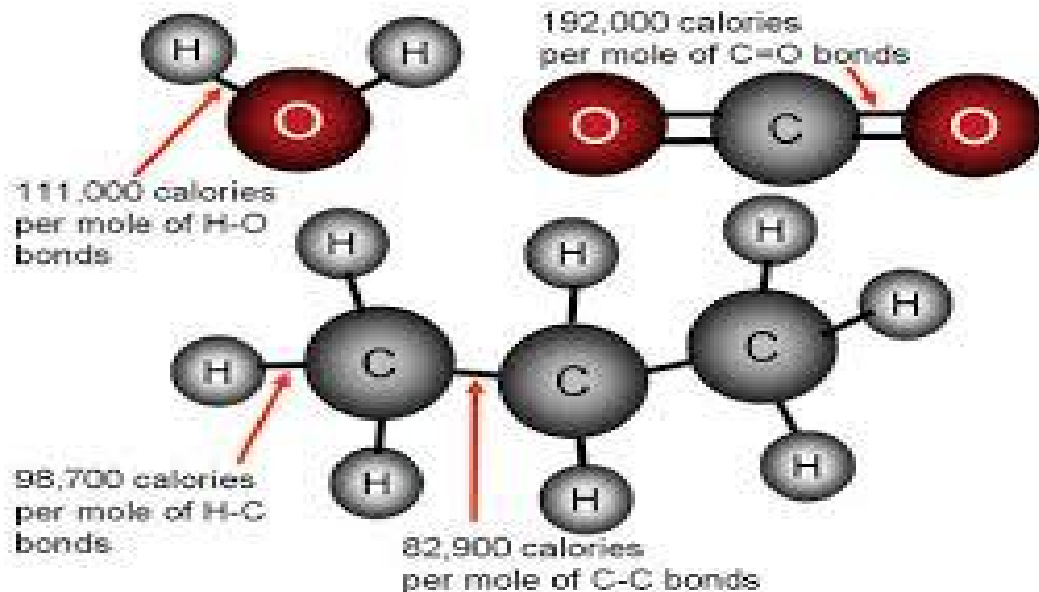
OR

Ag + Br  $\Rightarrow$  AgBr



# Breaking Chemical Bonds

- When chemical reactions occur, there is a change in energy
- Chemical bonds store energy
- When these bonds break, they release energy



# Heat

- If energy is released during a reaction, it is exothermic
- If energy is taken in during a reaction, it is endothermic



In an exothermic reaction, energy is released into the surroundings as heat. As a result, the temperature of the surroundings increases.



In an endothermic reaction, energy is absorbed from the surroundings. As a result, the temperature of the surroundings drops.

# Homework

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